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Biochemical limitations of *Bacillus thuringiensis* based biopesticides production in a wheat bran culture medium

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PII: S0923-2508(23)00018-9

DOI: <https://doi.org/10.1016/j.resmic.2023.104043>

Reference: RESMIC 104043

To appear in: *Research in Microbiology*

Received Date: 24 October 2022

Revised Date: 30 January 2023

Accepted Date: 31 January 2023

Please cite this article as: R. Barssoum, G. Al Kassis, R. Nasserredine, J. Saad, M. El Ghoul, J. Abboud, N. Fayad, S. Dupoiron, J. Cescut, .C.A. Aceves-Lara, L. Fillaudeau, M.K. Awad, Biochemical limitations of *Bacillus thuringiensis* based biopesticides production in a wheat bran culture medium, *Research in Microbiology*, <https://doi.org/10.1016/j.resmic.2023.104043>.

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22

23 Abstract

24 *Bacillus thuringiensis*, a gram-positive sporulating bacteria found in the environment,  
25 produces, during its sporulation phase, crystals responsible for its insecticidal activity,  
26 constituted of an assembly of pore-forming  $\delta$ -endotoxins. This has led to its use as a  
27 biopesticide, an eco-friendly alternative to harmful chemical pesticides. To minimize  
28 production cost, one endemic *Bacillus thuringiensis* sv. *kurstaki* (*Btk*) strain Lip, isolated from  
29 Lebanese soil, was cultivated in a wheat bran (WB) based medium (IPM-4-Citrus project EC n°  
30 734921). With the aim of studying the biochemical limitations of *Btk* biopesticide production  
31 in a wheat bran based medium, the WB was sieved into different granulometries, heat treated,  
32 inoculated with *Btk* Lip at flask scale, then filtered and separated into an insoluble and a  
33 permeate fractions. Several biochemical analyses, *ie.* bio performances, starch, elemental  
34 composition, total nitrogen and ashes, were then conducted on both fractions before and  
35 after culture. **On a morphological level, two populations were distinguished, the fine starch  
36 granules and the coarse lignocellulosic particles. The biochemical analyses showed that both  
37 the raw and sieved WB have a similar proteins content (0.115 g/gdm WB), water content  
38 (0.116 g/gdm WB) and elemental composition (carbon: 45 %, oxygen: 37 %, nitrogen: 3 %,   
39 hydrogen: 6 %, ashes: 5 %). The starch content was 17 %, 14 % and 34 % and the fermentable  
40 fraction was estimated to 32.1 %, 36.1 % and 51.1 % respectively for classes 2, 3 and 4. Both  
41 the elemental composition and Kjeldahl analyses showed that the nitrogen is the limiting  
42 nutrient of the culture.**

43 Keywords: *Bacillus thuringiensis*, wheat bran, elemental composition, fermentable fraction,  
44 biochemical limitations.

45

## 46 1 Introduction

47 Insects, be it plant pests or vectors of human diseases, continue to pose a worldwide problem  
48 in agriculture and public health [1]. To get rid of these insects, farmers have traditionally  
49 resorted to the excessive use of chemical pesticides, harmful to nature and human health [2].  
50 This has led the scientific community to look for less harmful alternatives, hence the  
51 introduction and industrialization of "biopesticides", mainly those produced by  
52 microorganisms [3]. *Bacillus thuringiensis* (*Bt*) is a facultative anaerobic gram-positive  
53 sporulating bacteria, that is among the most used in the production of biopesticides.  
54 Moreover, *Bt* has been a genetic source for the production of transgenic plants, able to secrete  
55 insecticidal molecules on their own [4]. It usually inhabits different environments such as soil,  
56 settled dust or water [5]. *Bt* has been shown to be toxic to various phytopathogen insects  
57 including lepidopterans, coleopterans, dipterans, or nematodes, but is considered safe for  
58 mammals [6]. *Bt* based products provide effective and eco-friendly pest control, be it disease  
59 vectors or phytopathogens problematic in agriculture and forestry. *Bt* life cycle has two main  
60 phases: (i) the exponential phase where the bacteria multiplies and produces vegetative  
61 biomass and (ii) the sporulating phase where *Bt* begins to sporulate [7] and produces crystals.  
62 The latter are a combination of Cry toxins, or  $\delta$ -endotoxins (<https://www.bpprc.org/>). These  
63 toxins are solubilized in the insect gut by the alkaline pH and activated by proteolysis. They  
64 will then interact with gut receptors to create pores in the intestinal cell membranes leading  
65 to the death of insect larvae [8]. **In addition, the *Bacillus cereus* group to which *Bacillus***  
66 ***thuringiensis* belongs, have a large number of transcriptional regulators, responsible for the**  
67 **activation of the biofilm formation [9].** *Bacillus thuringiensis* sv. *kurstaki* (*Btk*) is well known  
68 for its activity against lepidopteran larvae. Most of the biopesticides distributed in the world  
69 are mainly based on *Btk* HD1 [10]. Two endemic strains, *Btk* Lip [11] and Blb1 [12], isolated  
70 from Lebanese and Tunisian soils respectively, have exhibited a higher efficiency than HD1  
71 against the lepidopteran larvae *Ephestia kuehniella*.

72 For its growth and production, in the laboratory and on industrial scales, *Bt*, similarly to other  
73 bacteria, needs a source of sugar. Several sugars, including glucose, fructose, starch, maltose,  
74 and ribose, can be fermented by this bacterium [13]. Some new *Bt* strains have also been  
75 shown to ferment xylan and cellulose [14]. Glucose, in a limited quantity [15], is the  
76 bacterium's most utilized carbon source and the most effective for sporulation and  $\delta$ -

77 endotoxin synthesis [16]. Organic nitrogen is also a key element for bacterial growth. Some  
78 amino acids (leucine, valine, arginine) are crucial for biomass production and sporulation.  
79 However, other amino acids like cysteine have an opposite effect [17]. Minerals, like  
80 potassium ( $K_2HPO_4$ , 50-100 mM), are also required for  $\delta$ -endotoxins synthesis [18]. The  
81 Anderson medium (yeast extract, bactopectone, glucose and  $(NH_4)_2SO_2$ ) yielded  $7 \times 10^{11}$   
82 spores/mL compared to other media used in *Bt* culture [18,.19]. Özkan et al., 2003 found that  
83 NSYM (nutrient yeast salt medium) based on peptone, glucose, meat extract and yeast extract  
84 produced  $3 \times 10^{10}$  cfu/mL [21]. Because they contain yeast or animal extracts, all of these  
85 synthetic media were classified as complex media. Since the cost of production is a key step  
86 in biopesticide production at large scale, several other complex media with reduced cost have  
87 been proposed for *Btk* culture [22]. **Wheat bran (WB), a low- cost by-product (192 USD/  
88 tonnage) of cereal grain milling is estimated to be 650 million tons per year in the world  
89 [23], contains all the elements needed for the bacterial growth (sugar, proteins and  
90 minerals). As previously demonstrated, WB medium presented a higher yield for *Btk* crystals  
91 production compared to semi-synthetic medium [22, 23, 24]. Following an experimental  
92 design, a response surface methodology was employed to define the operating parameters  
93 for optimal *Bt* growth.** The wheat bran was sieved in order to understand the effect of the  
94 granulometry on biopesticide production.[27] In this context, IPM-4-Citrus project (MSCA  
95 RISE, n° 734921, April 2017-December 2022, [www.ipm-4-citrus.insa-toulouse.fr](http://www.ipm-4-citrus.insa-toulouse.fr)) developed  
96 simultaneously research and transfer activities about two new biopesticides based on the  $\delta$ -  
97 endotoxins produced by *Btk* Blb1 and Lip, active against citrus pests *Phyllocnitis citrella* and  
98 *Prays citri*. The project englobing 11 partners from 6 countries, aims to understand and raise  
99 awareness among stakeholders about the health risks related to citrus pests and to develop  
100 an alternative integrated pest management (IPM) approach based on biological control. These  
101 actions and scientific locks include bioproduction, formulation and bioactivity, and transfer to  
102 the market. To this end, the two selected strains of *Btk*, Blb1 and Lip and the one reference  
103 HD1, were cultivated in a wheat bran based medium (WB).

104 The aim of this study is to define the bacteria's nutritional requirements during the culture  
105 and to determine the biochemical limitations of the production process, such as the  
106 fermentable fraction and the limiting nutrient. To this end, the WB was sieved into three  
107 particle sizes (classes). Each class was suspended in water, sterilized by autoclaving and

108 inoculated by *Btk* Lip at flask scale. The suspensions were then filtered and separated into two  
109 fractions: the insoluble fraction containing the remaining WB and the permeate fraction  
110 containing the vegetative cells, the spores and the  $\delta$ -endotoxins. Several chemical and  
111 biochemical analyses were conducted on wheat bran, on the strains and on both fractions.

## 112 2 Materials and Methods

### 113 2.1 Strains and substrate

114 Three strains of *Bacillus thuringiensis sv. kurstaki* (*Btk*) were selected for this study:

- 115 1. *Btk* HD1, the industrial reference strain [28]
- 116 2. *Btk* Blb1, isolated from a Tunisian soil sample [12]
- 117 3. *Btk* Lip, isolated from a Lebanese soil sample [11]

118 Wheat bran (WB) as an industrial by-product is the main component of the culture medium  
119 to produce the *Btk* spores/crystals mixture. **The natural** WB was bought from Nehmet rabna-  
120 Grocery store (Mansourieh, Lebanon, 14/04/2021). This by-product was sieved and  
121 characterized by several physico-chemical analyses: dry matter, water content, minerals,  
122 protein, total sugar, nitrogen and elemental composition.

### 123 2.2 Sample preparation

#### 124 2.2.1 Sieving

125 Wheat bran (Nehmet Rabna, Mansourieh, Lebanon, 14/04/2021) was sieved in order to study  
126 its physicochemical characteristics. Batches (70 g) were sieved on a vibrating sifter (ELE  
127 international, UK, SN: 80-0352, max frequency: 60 Hz, width: 380 mm, height: 1085 mm during  
128 20 minutes at 5 Hz. Three square mesh sieves (250  $\mu$ m, 500  $\mu$ m, 850  $\mu$ m) generated 4 fractions  
129 (class 1 to 4). Material balances were established for 2 batches (200 g) and the biochemical  
130 composition are reported in table 1 with an average water content of 0.12 g/grams dry matter  
131 (gdm) and a density of 1.13 grams humid matter (ghm)/mL. Class 1 (>850 $\mu$ m) represents less  
132 than 1% w/w and is not of industrial interest considering the separation and formulation steps  
133 downstream of bioproduction. Classes 2, 3 and 4 were adopted for further biochemical and  
134 physical analyses.

### 135 2.2.2 Flask cultures

136 *Btk* strains were initially isolated on petri dishes containing T3 solid medium [29]. The dishes  
137 were incubated for 12 h at 30 °C. Transfer to liquid medium was ensured by inoculation of 6  
138 mL Luria Broth (LB) medium [30]. Each suspension was incubated for 12 h at 30 °C with shaking  
139 at 250 rpm. Cell growth was measured by optical density (OD) at 600 nm. **For the liquid**  
140 **fermentation**, 3.68 ghm of WB from classes 2, 3 and 4 were suspended in a total volume of  
141 50 mL in **500 ml flasks**, equivalent to a concentration of [WB]=73.6 ghm/L. **According to our**  
142 **previous study [27], this concentration was the optimal concentration for *Btk* growth.**  
143 **Moreover, on a physical level, the selected concentration does not represent any physical**  
144 **limitations because the interaction between the WB particles in this suspension is**  
145 **moderated. However, choosing a higher concentration means that the WB particles will**  
146 **have a stronger interaction with each other and this limited the oxygen transference to the**  
147 **culture which induces problems during the fermentation.**

148 Then, the suspensions were sterilized by autoclaving at 121 °C for 20 min. **It is of note that**  
149 **the autoclaving may generate some compounds that can inhibit the bacterial growth [31].**

150 Afterwards, the media were inoculated at an equivalent initial OD= 0.15 (600 nm) and  
151 incubated for 48 h at 30 °C with shaking at 250 **revolutions** per minute (rpm) and **an initial**  
152 **pH= 6.2.**

### 153 2.2.3 Solid/liquid separation

154 Flask contents (with or without culture) were filtered on a Whatman paper (#28413902, cut  
155 off: 13 µm), generating insoluble (substrate) and permeate (cells, spores, δ-endotoxins)  
156 fractions. This was only done for classes 2 and 3, not 4, because the fine particles of the latest  
157 clogged the filter. All chemical analyses were realized with both fractions for classes 2 and 3,  
158 and on the whole suspension for class 4, in order to establish the mass and elemental  
159 balances.

## 160 2.3 Physico-chemical analysis of the substrate

### 161 2.3.1 Morpho-granulometry of WB particles

162 The morphological analysis of the raw and sieved WB was performed ex-situ using a  
163 morphogranulometer (Mastersizer G3S, Malvern Instruments Ltd. SN: MAL1033756,  
164 Morphologi v7.21 software). This optical device includes a lens system (magnification: from

165 ×1 to ×50, min/max size: 0.5/3000 μm) and a camera (Nikon CFI60) with a resolution close to  
166 0.06 μm/pixel. The analyses were conducted in dry mode by dispersing WB (0.25 mg) through  
167 a specific unit (DSU) under a pressure of 4 bar for 10 ms. Image acquisition and analysis were  
168 performed according to a standard operating procedure (SOP) that defined the type of light  
169 source (diascopic light, bright mode), illumination parameters (light intensity: 80 % ± 0.2),  
170 magnification (×2.5) and particle detection threshold (thresholding = 140) and scanned area  
171 (40x40 mm). The number (raw data) and volume (conversion under assumption of a spherical  
172 model) distributions associated with each parameter (particle size, morphometry) were  
173 generated.

#### 174 2.3.2 Relative (x) and absolute (n) humidity of WB

175 In order to determine the dry WB moisture content, empty crucibles were dried in an oven  
176 (Thermo Fischer Scientific, ref: 0562202010) for 2 h at 105 °C and weighted. WB (3.68 ghm)  
177 was dried in the oven (24 h, 105 °C) then the total mass (crucible + WB) was measured.  
178 Absolute (g water/ gdm) and relative humidity (g water/ ghm) were then calculated:

$$179 \text{ humidity} = \frac{M_{\text{water}}}{\text{Dry or humid matter (WB)}} \quad \text{Eq. 1}$$

180

#### 181 2.3.3 Dry matter analysis

182 WB (3.68 ghm) was suspended in water (50 mL) then autoclaved (**20 min**, 121 °C, saturated  
183 steam sterilization). Insoluble and permeate fractions of sterilized suspensions were  
184 separated by filtration. The filter papers were first dried in an oven (2 h, 105 °C) and weighted  
185 before filtration and then they were put back in the oven for 24 h. The total filtrate was  
186 collected in 50 mL flasks. Empty crucibles were put in an oven for 2 h at 105 °C, followed by at  
187 least 1 h in the desiccator (SICO) until reaching a constant weight. Ten mL of permeate placed  
188 in crucibles were dried in an oven for 24 h at 105 °C. The dry matters of permeate (gdm sol  
189 WB/ g) and insoluble (gdm ins WB/g) fractions were measured and calculated [32].

#### 190 2.3.4 Ashes

191 Half a gram of insoluble fraction or 7.5 mL of the permeate were placed with the dried  
192 crucibles in the oven (600 °C, 2 h). The ashes concentration for both insoluble (g ash/gdm ins)  
193 and permeate (g ash/gdm sol) fractions was estimated. These analyses could help understand  
194 the mineral composition of insoluble and soluble fractions of the WB based medium.



### 195 2.3.5 Water retention capacity (WRC)

196 Empty Falcon tubes (50 mL) were first weighted, before adding 1.84 g of WB and suspending  
197 it in 25 mL of water. The suspensions were agitated for 60 min and sterilized in the autoclave  
198 (121 °C, 1 h) then vortexed (30 s) and centrifuged (3000 rpm, 15 min, 16 °C). The supernatant  
199 was eliminated and the pellet was weighted. The water retention capacity (% w/w) was then  
200 calculated [33].

## 201 2.4 Bioproduction analysis

### 202 2.4.1 Cells and Spores counting

203 First, dilutions **rate** of the permeate of a 48 h culture (30 °C) were realized (dilution rate: 10<sup>-1</sup>  
204 to 10<sup>-9</sup>). Then, petri dishes containing solid T3 medium were inoculated with 10 µL of diluted  
205 culture at 10<sup>-9</sup>, in triplicate, and then incubated at 30 °C for 12 h for counting of vegetative  
206 cells and living spores. As described for total biomass, spore concentrations alone were  
207 counted by inoculating diluted culture on T3 medium, after actively killing vegetative cells by  
208 heat treatment (80 °C, 10 minutes). Spores and cfu concentrations were then calculated  
209 considering the average count of the triplicate (A), the dilution factor (F<sub>d</sub>) and the inoculum  
210 volume (v):

$$211 \text{ [Total flora] or [Spores] } = \frac{A \cdot F_d}{v} \left( \frac{\text{CFU or Spores}}{\text{mL}} \right) \quad \text{Eq. 2}$$

### 212 2.4.2 Protein dosage (Bradford)

213 Protein concentration was estimated using the Bradford reagent [34]. After 48 h of culture, 1  
214 mL of spores/crystals mixture were centrifuged (13000 rpm, 16 °C, 5 min). The supernatant  
215 was thrown out and the pellet was washed twice with NaCl 1M/Triton 0.01 M and four times  
216 with 1 mL of cold autoclaved water. The crystals in the pellet were solubilized with 50 mM of  
217 NaOH by incubating at 30 °C for 2 h in a rotary shaker (250 rpm). **The samples were**  
218 **centrifugated and 10 µl of the supernatant were mixed with 200 µl of Bradford reagent, and**  
219 **790 µl of water [31].** The corresponding tubes were incubated in the dark for 12 min. The  
220 optical density was then measured at λ = 595 nm, and the protein concentration in the  
221 samples was calculated based on a standard calibration curve established using BSA  
222 standards.

## 223 2.5 Biochemical analysis

### 224 2.5.1 Determination of starch content

225 Starch is an important source of glucose for the bacterial culture. To determine the starch  
226 content in WB, the colorimetric method [30, 31] was used. Iodine solution was prepared by  
227 crushing 22 mg of Iodine and mixing it with 10 mL of water. KI (44 mg) was added and the total  
228 volume was adjusted to 50 mL and homogenized. The standard range was prepared by using  
229 starch solutions (Commercial ref: S58150-3J - batch number: 19005-25-81) at concentrations  
230 varying between 0.05 and 3 g/L. Calibration curve was constructed by adding 0.2 mL of a  
231 standard solution with 1 mL of I<sub>2</sub>/KI. The solution was left to react for 2 min at room  
232 temperature and the optical density was measured at 540 nm (spectrophotometer: Perkin  
233 Elmer, ENSPIRE, SN 23000751). Starch was quantified by determining the dilute solutions  
234 absorbance using the Beer-Lambert law considering that  $\epsilon$  is the slope of the standard curve:

$$235 \text{Abs}_{\lambda} = \epsilon_{\lambda} \times l \times c = 0.825 \times l \times c \quad \text{Eq. 3}$$

236

### 237 2.5.2 Elemental composition: CHONS

238 The elemental composition analysis [37] was conducted **at the beginning and the end of the**  
239 **culture** using the Flash Smart 2000 (Thermo Fischer Scientific, ref: 20140222). The analysis is  
240 based on flash combustion (1800 °C) under an inert atmosphere (Helium) followed by gas  
241 chromatography. Analysis of CHONS were done through two distinct analysers. The machine  
242 is constituted of a quartz tube (Thermo Fischer Scientific, ref: 46802015) and two gas  
243 chromatography columns (ref: 26007900 for CHNS, ref: 26008215 for O) and a thermal  
244 conductivity detector with increased sensibility. The experiment was done under the following  
245 conditions: helium flow rate: 950 mL/min, oven temperature: 950 °C and pressure below 5  
246 kPa.

247 Sample preparation included a preliminary step to remove residual water. It is required to  
248 ensure that hydrogen and oxygen only come from the substrate or cell activity. For the  
249 insoluble fraction, frozen solid samples were dried in an oven (24 h, 70 °C, 300 mbar) whereas  
250 for the permeate, liquid samples were thawed then lyophilized (Freeze drier: Epsilon 2-4 LSC  
251 plus, ref: TWB /MO/PS/2, 19/10/2017) for 48 h under a pressure of 10 mbar.

252 Dried samples were weighted on a precision balance (METTLER TOLEDO, Model XP6/52, SNR  
253 B444194554, precision: 0.1  $\mu\text{g}$ , maximum capacity: 2.1 g) in Sn capsules (Thermo Fischer  
254 scientific, ref: 24006400) for C, H, N and S analysis and Ag (thermo Fischer scientific, ref:  
255 24005400) for oxygen analysis. The masses (sample, capsule) ranged between 2 and 3 mg. The  
256 capsules were rolled into balls with pliers, then placed on a carousel for analysis. All samples  
257 were analysed in duplicate.

258 Calibrating the device with a standard is a critical step. The standard, bbot (ref: PJ. 33835210),  
259 has a known mass elemental composition: C (72.53 %), H (6.09 %), O (7.43 %) N (6.51%) and S  
260 (7.44 %). The standard range consisted of 5 samples of different masses between 2 mg and 4  
261 mg. The calibrating curves were established. The equations, precision and correlation  
262 coefficient are shown in the supplementary material.

263 All analyses with an area below or above the limits of detection (lods) will be considered  
264 outside of the calibration range. Nevertheless, since linear regression passes through the  
265 ordinate at the origin, the weak values will be interpreted (supplementary material).

266 Statistical analysis (mean and standard deviation) of CHONS analysis was systematically  
267 reported. The statistical inference to compare samples is carried out by a Student test (t-test).  
268 Assuming a Normal distribution and considering that the variables are quantitative and  
269 discrete and that samples are independent, the variable t allowed to compare 2 samples for a  
270 population size  $n < 30$ . The analysis of variance and mean is performed using the t-test  
271 function (Microsoft Office Excel, 2019), the samples were compared item by item with a p-  
272 value  $< 0.05$ .

273 2.5.3 Nitrogen (Kjeldhal): Ntot, Nmin, Norg, Eq. protein

274 The Kjeldahl method [38] was applied to quantify total, mineral and organic nitrogen.  
275 Insoluble and permeate fractions for sterilized WB before and after culture were analysed.  
276 The evolution of nitrogen during culture can be interpreted to evaluate cell growth,  
277 sporulation,  $\delta$ - endotoxins production and nitrogen consumption.

278 Kjeldahl mineral nitrogen (KMN) was quantified by adding 0.5 g of the insoluble fraction (dried  
279 WB) or the permeate (7.5 mL) with 10 mL of sulfuric acid (Sigma-Aldrich, ref: 1506921000,  
280 molarity: 98 %). The mixture was heated to 350 °C in a heating block. The tube was placed in  
281 the Kjeldahl machine (Kjeltec™, 8400). 25 mL of boric acid (Sigma-Aldrich, ref: B6768,

282 molarity: 4 %) were added in an Erlenmeyer with: 40 mL NaOH (Sigma-Aldrich, ref:  
 283 1587931000) and 40 mL distilled water for 4 min. The Tashiro indicator was added to the  
 284 Erlenmeyer content and green colour was obtained. A titration with sulfuric acid solution (0.05  
 285 N) was done to obtain the pink colour. Each mole of ammonium corresponds to 2 moles of  
 286 sulfuric acid. The number of moles of nitrogen is equal to the ammonium one. To deduce the  
 287 mass of nitrogen, the number of moles was converted to mass by considering the nitrogen  
 288 molar mass (14 g/mol) hence the factor 0.014 was deduced. The mineral nitrogen  
 289 concentration was calculated as follows:

$$290 \quad KMN = \frac{V_{H_2SO_4} * N * 0.014007}{m_{sample}} \left( \frac{g \text{ Nmin}}{g \text{ dm}} \right) \quad \text{Eq. 4}$$

291 Kjeldahl total nitrogen (KTN) was quantified by adding 0.5 g of the insoluble fraction (WB) or  
 292 7.5 mL of the permeate in Kjeldahl tubes with 10 mL of sulfuric acid (98%), 5 g of Kjeldahl  
 293 catalyst and 3 glass beads. The tubes were incubated in the mineralizer for 2 h, then each  
 294 sample was passed throughout the Kjeldahl machine (same conditions of the mineral  
 295 nitrogen). The total nitrogen concentration was calculated as follows:

$$296 \quad KTN = \frac{V_{H_2SO_4} * N * 0.014007}{m_{sample}} \left( \frac{g \text{ Ntot}}{g \text{ dm}} \right) \quad \text{Eq. 5}$$

297 The concentration of organic nitrogen was then determined:

$$298 \quad [Norg] = [KTN] - [KMN] \quad \left( \frac{g \text{ Norg}}{g \text{ DM}} \right) \quad \text{Eq. 6}$$

299 The proteins amount was deduced by considering a vegetative factor  $F = 6.25$  g Protein Bovine  
 300 Serum Albumin (BSA)/ g nitrogen:

$$301 \quad [proteins] = [Norg] * F \quad \left( \frac{g \text{ proteins}}{g \text{ DM}} \right) \quad \text{Eq. 7}$$

302 The analysis of variance and mean is performed using the T-test function (Microsoft Office  
 303 Excel, 2019), the samples were compared item by item with a p-value < 0.05.

## 304 3 Results

### 305 3.1 Substrate characterization

#### 306 3.1.1 Sieving and granulometry

307 Firstly, to characterize the WB itself and the different classes, the raw material was sieved.  
 308 Four classes were generated: class 1 >850  $\mu\text{m}$ , class 2 500-850  $\mu\text{m}$ , class 3 250-500  $\mu\text{m}$ , class  
 309 4 <250  $\mu\text{m}$ . The average post-sieving material balance of the substrate indicates that class 3 is  
 310 dominant (60.5 % w/w) followed by classes 2 (19.9 % w/w) and 4 (19 % w/w). The particle size

311 analysis of the dry substrate allowed the establishment of distribution functions (in number  
312 and volume) of the particles as a function of the equivalent circle diameter ( $d_{CE}$ ). Comparative  
313 analysis of distribution functions combines the use of graphical representations (such as  
314 dendrograms) and descriptive statistics (e.g., mean, median, mode and standard deviation).  
315 The assessment of normality can be based on the comparison between the sample moments  
316 and the theoretical moments of the normal distribution (method of "moments"). The  
317 similarities between the distribution functions can be examined, by applying the Student's t-  
318 test and determining the overlap coefficient [35, 36].

319 Two populations, "fines" and "coarse" are present. Morphological analysis of the particles  
320 indicates that the "fines" have an ovoid shape (aspect ratio= 0.735, circularity = 0.849) with a  
321 smooth surface and that the granules can be aggregated. The "coarse" ones have a rough  
322 surface (aspect ratio = 0.700, circularity = 0.783) and a rather angular polyhedral geometry  
323 (with 3, 4 or 5 sides) in coherence with the milling process and the extraction of the pericarp  
324 of the wheat grain by mechanical attrition following a wetting operation. Considering the  
325 deviation of these morphologies from the sphere model, the transition from a number  
326 distribution to a volume distribution can be considered. The number,  $En(d_{CE})$  and volume,  
327  $Ev(d_{CE})$  distributions highlight the ratio of "fines" (5-50  $\mu\text{m}$ ) and "coarse" (150-1000  $\mu\text{m}$ )  
328 population of granules. Figure 1 illustrates the bimodal nature of each class. The volume  
329 distribution and the circle diameter equivalent to 90 % of the population, ( $Dv0.9$ ),  
330 demonstrate that the sieving effect on coarse particles agrees with the maximum size of a  
331 square mesh (1200, 700 and 350  $\mu\text{m}$ ). Nevertheless, the volume distributions show a wide  
332 range ( $Dv0.1$  to  $Dv0.9$ ) and a low overlap rate (obtained by calculating the integral of the  
333 distribution function) of 25.68 % (between classes 2 and 3) and 29.71 % (between classes 3  
334 and 4). At this stage a secondary population between 5 and 50  $\mu\text{m}$  seems to be present but  
335 negligible in volume.

336 3.1.2 Dry matter, water content ( $n, x$ ) and biochemical composition of WB substrates

337 To explore the impact of granulometry on the biochemical composition of WB classes, water,  
338 starch, proteins, ashes contents and elemental composition were studied for both raw and  
339 sieved WB. All data was summarized in table 1.

340 No difference was observed in the water content neither in the raw WB nor in that sieved to  
341 classes (0.116 g water/gdm). Similarly, the ash content varies slightly from 3.23 % to 4.43 %

342 ( $p > 0.05$ ). The starch content of WB varies inversely to the particle size: 0.173 g/gdm (class 2),  
343 0.144 g/gdm (class 3) and 0.347 g/gdm (class 4). The Kjeldahl analysis demonstrated that the  
344 total nitrogen is equal to 0.025, 0.027 and 0.03 g/gdm WB respectively for classes 2, 3 and 4.  
345 On the other side, the organic nitrogen was calculated to be: 0.018, 0.020 and 0.025 g/gdm  
346 WB for classes 2, 3 and 4. Protein content was then deduced from organic nitrogen content  
347 (vegetative factor: 6.25) and was estimated to be 0.127, 0.145 and 0.155 g/gdm WB for classes  
348 2, 3 and 4. However, these differences were not significant. The mean elemental  
349 composition of the substrate is 44.37 % ( $\pm 0.37$ ), 6.59 % ( $\pm 0.17$ ), 36.6 % ( $\pm 0.42$ ) and 2.53 %  
350 ( $\pm 0.2$ ) for carbon, hydrogen, oxygen, and nitrogen, respectively. Sulfur remains an  
351 unquantifiable trace element with this method of analysis (limit of quantification  $loq < 0,5$  %  
352 w/w) regardless of its class. The differences in elemental composition between the different  
353 particle sizes are not significant (Student test,  $p > 0.05$ ). Based on the mean value of elemental  
354 composition, the mass and molar formula of WB were calculated to be:  $CH_{0.14}O_{0.71}N_{0.05}ash_{0.1}$   
355 and  $CH_{1.74}O_{0.53}N_{0.04}$  respectively.

356 In addition, a physical property, the water retention capacity, (WRC) was included in this  
357 study. The variation of the WRC decreases with the decrease of the particle size: 500% w/w  
358 for class 2, 400 % w/w for class 3 and 250 % w/w for class 4.

### 359 3.2 Bioproduction and fermentable fraction.

360 Following, to better understand the evolution of the WB medium composition after Lip  
361 culture, we studied subsequently the elemental composition of the *Btk* strains.

#### 362 3.2.1 Comparison between strains and WB elemental composition:

363 The elemental composition of the strains (Lip, Blb1 and HD1) following culture in LB was  
364 characterized and was shown to be identical as represented in table 3. Based on the mean  
365 value of the elemental composition, the strains mass and molar formula were calculated to  
366 be:  $CH_{0.14}O_{0.59}N_{0.24}ash_{0.15}$  and  $CH_{1.72}O_{0.44}N_{0.21}$  respectively. The elemental composition of the  
367 strains was then compared to literature as shown in table 3. *Btk* strains elemental composition  
368 come closest to *Bacillus cereus*. Hereafter, a comparison was established between the mean  
369 elemental composition of the strains and the WB one. As highlighted in Fig.2, the strains are  
370 less rich in oxygen, but richer in nitrogen compared to the WB ( $p < 0.05$ ). This difference is also  
371 confirmed by comparing the strain and the WB molar formula. **The observed difference**  
372 **between the nitrogen composition of our *Btk* strains and that of WB cannot be definitively**

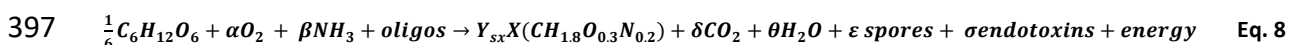
373 **explained. Nonetheless, a speculation could be made about this difference: it is due to**  
374 **bacteria being quite rich in proteins, be it metabolic proteins, enzymes, structural proteins,**  
375 **surface receptors or others, hence their richness in nitrogen, a major component of amino**  
376 **acids.**

### 377 3.2.2 Cell and spore counting and $\delta$ -endotoxin production

378 To determine the fermentable fraction, it is important to assess the productivity of *Btk* Lip ( $\delta$ -  
379 endotoxins concentration), and the distribution of vegetative cells and living spores in the  
380 permeate. After 48 hours of culture produced in various classes of WB media, the  
381 concentration of spores and  $\delta$ -endotoxins was examined. No significant difference of spore's  
382 concentration was observed among classes. As for the  $\delta$ -endotoxins yield, a small difference  
383 was noted with class 2 and 3, compared to class 4. However, protein quantification was not  
384 significantly different (table 2).

### 385 3.2.3 Mass balance and fermentable fractions

386 To estimate the fermentable proportion across classes, the mass balance between insoluble  
387 and permeate fractions was established (Fig. 3). Our findings revealed that, after culture, the  
388 mass balance of the insoluble fraction decreased from 0.85 g/gdm WB to 0.63 g/gdm WB.  
389 However, the mass balance in the permeate increased from 0.15 g/gdm WB to 0.25 g/gdm  
390 WB for both classes 2 and 3. Due to a filter clogging induced by fine particles with WB class 4,  
391 the separation was not possible and the mass balance analysis for this class corresponded to  
392 both insoluble and permeate fractions. Assuming the stoichiometry of the bioreaction that  
393 describes the growth and the sporulation, and with a hypothesis that the production yield  
394  $Y_{s/x} = 0.5$ , the fermentable fraction was calculated to be 32.3, 36.1 and 51.1 % for classes 2, 3  
395 and 4 respectively which is far from WB's starch content of each class (17.3 % (class 2), 14.4 %  
396 (class 3) and 34.7 % (class 4)) (table 1).



### 398 3.2.4 Elemental composition (CHONS)

399 Knowing the elemental composition of the strains and the WB, and having determined the  
400 fermentable fraction among WB classes, the next step was to know how the WB elemental  
401 composition would evolve and what would be the limiting nutrient. Our results demonstrated  
402 that in the insoluble and permeate fraction, and for both classes 2 and 3, the carbon,  
403 hydrogen, ashes and oxygen contents did not represent any significant difference ( $p > 0.05$ )

404 before and after culture. However, for class2, nitrogen significantly, decreased ( $p < 0.05$ ) from  
405 2.67 % to 1.49 % in the insoluble fraction and increased ( $p < 0.05$ ) from 1.44 % to 5.58 % in  
406 permeate one. Similar pattern was observed for class 3. Minerals also increased in permeate,  
407 approximatively from 5 % to 10 % in both classes 2 and 3 (Fig. 4).

#### 408 3.2.5 Nitrogen and Proteins (Kjeldahl method)

409 Since nitrogen concentrations in the insoluble and the permeate fractions significantly  
410 changed after culture, a more thorough analysis of the nitrogen was conducted. Hence, the  
411 nitrogen concentration was examined using the Kjeldahl method in order to understand the  
412 evolution of organic, mineral, and total nitrogen in both fractions. For classes 2 and 3, after  
413 culture, the total nitrogen significantly decreased from 0.03 g/gdm WB to 0.02 g/gdm WB ( $p <$   
414  $0.05$ ) in the insoluble fraction, along with the significative decrease ( $p < 0.05$ ) of the organic  
415 nitrogen (0.025 g/gdm WB to 0.02 g/gdm WB). The mineral nitrogen remained stable at 0.01  
416 g/gdm WB (Fig. 5A). In the permeate fraction, for classes 2 and 3, the total nitrogen increased  
417 significantly from 0.025 to 0.075 g/gdm WB ( $p < 0.05$ ) along with the increase ( $p < 0.05$ ) of the  
418 organic nitrogen from 0.015 g/gdm WB to 0.035 g/gdm WB for class 2 and 0.05 g/gdm WB for  
419 class 3. Even though the mineral nitrogen increased from 0.01 g/gdm WB to 0.025 g/gdm WB  
420 in the permeate (Fig. 5B) but this increase was not significant ( $p > 0.05$ ). Thus, the ratio mineral  
421 nitrogen/organic nitrogen decreases after culture in the permeate fraction and increases in  
422 the insoluble fraction. Considering class 4, and due to its tiny particles size, there was no  
423 separation into fractions. However, the total nitrogen in this suspension increased significantly  
424 ( $p < 0.05$ ) from 0.03 to 0.05 g/gdm WB along with the organic nitrogen (0.025 to 0.04 g/gdm  
425 WB). These results are consistent with the permeate results for classes 2 and 3, thus we  
426 showed them as a permeate in Fig.5B.

#### 427 4 Discussion

428 Producing low cost biopesticides is one of the main objectives of IPM-4-Citrus project. Present  
429 work explores more precisely: how WB media satisfies the nutritional requirements of the  
430 strains, the evolution of the biochemical composition during the culture and the identification  
431 of the limiting nutrient. Several scientific and technical bottlenecks were cleared up by: (i) size  
432 distribution of WB substrate after sieving and establishing the mass balance, (ii) quantifying  
433 the fermentable fraction, (iii) determining the biochemical composition of both the WB and



434 the strains and (iv) identifying the limiting nutritional factors by an analysis of elemental  
435 composition.

436 The granulometry of the WB particles affect the downstream process (DSP) during the  
437 clarification of the fermentation broth, the solid/liquid separation and the formulation  
438 process (spray drying). Moreover, after formulation, remaining coarse particles may plug the  
439 nozzles during field treatment and spreading. Therefore, a subsequent morpho-granulometry  
440 analysis was used to identify the WB particle range composing each class. On a morphological  
441 level, the spherical shape of the fine particles may be attributed to starch granules. On one  
442 hand, the number distribution functions confirm that the fine population is dominant. This  
443 population is concentrated between 5 and 50  $\mu\text{m}$  particle size, with characteristic values that  
444 are more or less identical for the three classes. On another hand volume distribution did  
445 confirm the effect of sieving on the coarse particles. However, nor number, nor volume  
446 distribution were enough to estimate the relative proportion of 'fines' and 'coarse' particle  
447 populations. This is due to the fact that starch granules remain attached to lignocellulosic  
448 fraction. Therefore, the biochemical studies were carried out in order to analyse the WB starch  
449 content, and the results showed that it was 0.173, 0.144, and 0.347 g/gdm WB for classes 2,  
450 3, and 4, respectively. This result confirm that the dominant proportion of starch is found in  
451 class 4. Wet processing and heating (sterilization) of the culture media prior to inoculation  
452 improved the accessibility of the starch to the microorganism by inducing swelling and starch  
453 release into the suspended wheat bran [41].

454 Hereafter, an enlarged chemical analysis of raw and sieved WB was conducted. Our results  
455 were consistent with those previously reported in the literature [36, 37, 38, 39, 40, 41, 42]  
456 proving that the starch content in WB varies between 0.13 and 0.40 g/gdm WB among classes.  
457 The protein (around 0.15 g proteins/gdm WB) and the water content (between 0.015 and  
458 0.019 g water/gdm WB) and the elemental composition of the WB were shown to be  
459 independent from the particle size. Taking into consideration all the obtained results regarding  
460 the chemical analysis of raw and sieved WB samples, the total mass balance is supposed to be  
461 100 %. However, it was found to be slightly inferior (95.54 % for raw WB, 95.58 % for class 2,  
462 94 % for class 3, and 95.52 % for class 4), which was caused by the experimental error and the  
463 fact that the trace elements (phosphor, zinc, calcium, etc.) were not quantified.

464 Moreover, our result showed that the water retention capacity (WRC, explained by the ability  
465 of a matrix to trap water), was proportional to the particle size. The class 2 (highest  
466 granulometry) has the biggest capacity to retain water [49]. This is due to its richness in  
467 lignocellulosic matrix and its low starch content. Additionally, WRC decreases in class 4  
468 prepared suspension. This can be explained by the fact that starch granules aggregate and  
469 prevent the matrix from retaining water [50].

470 After confirming the productivity (spores and  $\delta$ -endotoxins concentration of *Btk*-Lip) in the  
471 different classes of WB-based media, we focused on fermentable fraction estimation across  
472 classes by evaluation the mass balance in insoluble and permeate fractions relative to WB dry  
473 matter content. Following culture, the mass balance in the insoluble fraction decreased for  
474 classes 2 and 3. However, it increased in the permeate. In addition, in total (insoluble+  
475 permeate), a decrease in the mass balance can be noticed. Based on the general fermentation  
476 reaction, this decrease is equal to the half of the fermentable fraction since the other half is  
477 lost as carbon dioxide. Knowing that glucose is the preferred source of carbon for *Btk* growth  
478 [19], we wondered whether the fermentable fraction is equal to starch content. Based on our  
479 results, starch concentration was much lower than the calculated fermentable fraction (starch  
480 concentration in WB equal approximatively 50% of the fermentable fraction). Based on this  
481 information, we deduced that *Btk* have used an alternative source of carbon to grow. This  
482 source is most probably the lignocellulosic matrix of the WB, **since *Btk* genetically is capable**  
483 **to novel cellulases which could liberate glucose from lignocellulosic matrix.**

484 In order to better understand and evaluate more deeply, the nutrient requirements of  
485 bacteria growing in a wheat bran-based medium, the elemental composition analyses were  
486 conducted first on the WB, then on the strain and lastly on the medium after *Btk* culture for  
487 48 hours [51]. Our results show that, independently from the strain origin, the elemental  
488 composition of all *Btk* strains is similar and consistent with what has been described in the  
489 literature particularly the one of *Bacillus cereus* [45, 46, 47]. Furthermore, the comparison  
490 between the strains and the WB elemental composition affirms that, the strains are  
491 particularly richer in nitrogen than the WB. Since no supplements were added to the medium,  
492 the WB is the only nitrogen source during *Btk* growth and sporulation. Consequently, this  
493 information was essential for identifying a component to track the change in the chemical  
494 composition of the WB during the fermentation process. Moreover, to determine the culture

495 limiting nutrient, we asked how the elemental composition of the WB would evolve across  
496 classes in both insoluble and permeate fraction. Five elements were analyzed: carbon, oxygen,  
497 hydrogen, nitrogen and minerals. After culture, in the insoluble fraction, only nitrogen  
498 decreased significantly. However, in the permeate fraction, nitrogen increased significantly.  
499 In addition, we can observe in the insoluble fraction, that only 30 % of the available nitrogen  
500 was consumed which affirms that the residual nitrogen is inaccessible for the bacteria from a  
501 physical standpoint. These results confirm that nitrogen is the culture limiting nutrient that  
502 induces the spores/crystals formation.

503 Afterwards, since the nitrogen could emanate from both mineral or organic components of  
504 WB, we wanted to assess which type of nitrogen *Btk* has consumed. To this end, we analyzed  
505 the variation of the total, the mineral and the organic nitrogen in both insoluble and permeate  
506 fractions. Our results did show that in the insoluble fraction, the total nitrogen decreased  
507 significantly along with the decrease of the organic nitrogen. However, in the permeate  
508 fractions, the total and the organic nitrogen increased significantly. The mineral nitrogen, on  
509 the other hand, did not represent any significant changes. Since the organic nitrogen is in  
510 principle found in the proteins, we can deduce, that *Btk* has partially consumed the proteins  
511 of the WB. This information affirms that sugar is not the only ingredient required for *Btk*  
512 growth and sporulation. Furthermore, as previously described, the strain is richer in nitrogen  
513 compared to the WB. This can explain the increase of the organic nitrogen in the permeate  
514 that is probably due to the production of the biomass. These results obtained by Kjeldahl are  
515 aligned with those obtained by elemental composition technique.

516 Based on all these results, we can conclude that the WB stands as an interesting by-product  
517 for biopesticide production since its composition meets the full nutritional requirements of  
518 *Btk* strains with nitrogen being the limiting factor. However, in order to optimize  
519 bioproduction on a large scale, the analysis should be conducted on a bioreactor level.  
520 Moreover, to improve the appropriate downstream process for biopesticides formulation and  
521 application, the physical limitations of such a medium should be examined.

## 522 **Conflict of interest**

523 The authors declared no conflict of interest.

## 524 5 Funding

525 EU Project IPM-4-CITRUS - Horizon 2020 – <https://www.ipm-4-citrus.insa-toulouse.fr> Grant  
526 n°734921 (April 2017- January 2023). Agence Universitaire de la Francophonie AUF- Bureau  
527 du Moyen Orient, BMO (November 2021) and INSA funds (Septembre 2022) (Mobility of  
528 Mireille KALLASSY from Lebanon to INSA Toulouse)

529

## 530 6 Acknowledgements

531 A special thanks to the organizing committee of the 10<sup>th</sup> international conference on *Bacillus*  
532 *cereus*, *Bacillus anthracis* and *Bacillus thuringiensis* (Paris, April 2022).

533

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## 709 8 Legends for tables

710

711 Table 1: Wheat bran characterization per class

712 Table 2: *Btk* Lip Spore counting and protein content with classes 2, 3 and 4713 Table 3: Elemental composition (% g/gdm), molar composition (g/c\_mol) of biomass and molar mass of biomass (g/c-mol)  
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Table 1: Wheat bran characterization per class

Class	Size ( $\mu\text{m}$ )	Granulometry ( $\mu\text{m}$ )				Mass balance (mean value)	Water content [g water /gdm]	WRC % g/gdm	Starch [g/gdm]	Total nitrogen [g/gdm WB]	Organic nitrogen [g/gdm WB]	Protein KJ [g/gdm]	Elemental composition %				
		Dv <sub>0.1</sub>	Dv <sub>0.5</sub>	Dv <sub>0.9</sub>	D(4,3)								[%w/w]	C	H	O	N
Raw		270.6	557.4	969.5	598.5	100	0.119 $\pm 0.07$		0.207 $\pm 0.008$				44.79 $\pm 0.44$	6.74 $\pm 0.20$	36.53 $\pm 0.37$	2.50 $\pm 0.24$	4.43 $\pm 0.12$
2	500-850	687.8	852.1	1055.8	865.4	19.80	0.115 $\pm 0.04$	500	0.173 $\pm 0.35$	0.025 $\pm 0.01$	0.018 $\pm 0.008$	0.127 $\pm 0.012$	44.21 $\pm 0.19$	6.50 $\pm 0.12$	35.62 $\pm 0.45$	2.60 $\pm 0.23$	3.70 $\pm 0.12$
3	250-500	325.1	531.3	761.5	531.1	51.41	0.116 $\pm 0.04$	400	0.144 $\pm 1.35$	0.027 $\pm 0.01$	0.02 $\pm 0.01$	0.145 $\pm 0.014$	45.34 $\pm 0.08$	6.46 $\pm 0.04$	37.89 $\pm 0.44$	2.48 $\pm 0.04$	4.28 $\pm 0.12$
4	<250	166.0	269.8	391.1	275.3	28.20	0.116 $\pm 0.04$	250	0.347 $\pm 0.47$	0.03 $\pm 0.02$	0.025 $\pm 0.01$	0.155 $\pm 0.016$	42.40 $\pm 0.04$	6.61 $\pm 0.12$	34.60 $\pm 0.37$	2.61 $\pm 0.24$	3.23 $\pm 0.12$

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720

Table 2: *Btk* Lip Spore counting and protein content with classes 2, 3 and 4

Class	Spore [cfu/mL]	Endotoxin [g eq. BSA/L]
2	3.50E+09 ± 2.36E+08	0.549 ± 0.117
3	4.12E+09 ± 1.63E+09	0.547 ± 0.075
4	3.50E+09 ± 1.04E+09	0.432 ± 0.062

721

722 Table 3: Elemental composition (% g/gdm), molar composition (g/C\_mol) of biomass and  
723 molar mass of biomass (g/C\_mol) obtained from experimental data (*Btk*) and literature

724 \*na =not analysed

Strain	Elements (% w/w)					Molar formula	Molar Mass
	Ashes	C	H	O	N	(C_mol)	(g/C_mol)
<i>Btk</i> Blb1*	7	43.37	6.22	24.84	10.48	CH <sub>1.72</sub> O <sub>0.45</sub> N <sub>0.21</sub>	25.98
<i>Btk</i> HD1	6.7	45.7 ±0.067	6.61 ±0.090	26.89 ±0.870	10.51 ±0.011	CH <sub>1.7</sub> O <sub>0.42</sub> N <sub>0.2</sub>	25.42
<i>Btk</i> Lip	na	44.65 ±0.300	6.42 ±0.072	26.56 ±0.680	11.23 ±0.068	CH <sub>1.73</sub> O <sub>0.43</sub> N <sub>0.19</sub>	25.51
<i>Btk</i> - mean	6.80	44.82 ±0.330	6.46 ±0.117	26.35 ±0.775	10.79 ±0.039	CH <sub>1.72</sub> O <sub>0.44</sub> N <sub>0.21</sub>	25.52
<i>Lactobacillus</i> <i>helveticus</i> (Fagerbakke et al, 1996)	9.03	47.54	6.25	24.39	12.79	CH <sub>1.58</sub> O <sub>0.39</sub> N <sub>0.23</sub>	25.26

<b><i>Flavobacterium Dehydrogenans</i></b> <i>(Lawford Rousseau et al, 1996)</i>	13.5	45.16	6.15	24.29	10.87	CH <sub>1.63</sub> O <sub>0.40</sub> N <sub>0.21</sub>	26.60
<b><i>Escherichia coli</i></b> <i>(Popovic et al, 2019)</i>	11.3	47.83	6.95	21.65	12.3	CH <sub>1.74</sub> O <sub>0.34</sub> N <sub>0.22</sub>	25.11
<b><i>Bacillus cereus</i></b> <i>(Popovic et al, 2019)</i>	9.98	46.05	5.73	26.26	11.98	CH <sub>1.49</sub> O <sub>0.43</sub> N <sub>0.22</sub>	26.08

725 na: not analysed

726 \*Blb1 elemental composition analysis was conducted once due to a lack of material

## 727 9 Supplementary material

728 Precision and calibration coefficients for CHONS analysis

Element	Element mass (for 2 mg BBOT)	Equation ( $m = a \times \text{signal}$ )	Precision (%)	R <sup>2</sup>	LOD min ( $\text{timex}\Delta\theta$ )	LOD max ( $\text{timex}\Delta\theta$ )
C	1,45	$a = 323 \times 10^7$	2,190	0,997	$7 \times 10^6$	$17 \times 10^6$
H	0,12	$a = 890 \times 10^7$	2,240	0,999	$19 \times 10^6$	$48 \times 10^6$
N	0,14	$a = 132 \times 10^7$	2,030	0,999	$26 \times 10^4$	$70 \times 10^4$
S	0,15	$a = 116 \times 10^7$	10,530	0,918	$19 \times 10^4$	$63 \times 10^7$
O	0,15	$a = 231 \times 10^7$	3,300	0,986	$47 \times 10^7$	$10 \times 10^6$

729

Figure legends

Fig 1: Cumulative distribution function in number and volume for WB particles (classes 2, 3 and 4) including a microscopic image of WB class2 (Diascopic illumination, Intensity 80%, surface length: 10x10 mm<sup>2</sup>, BF, magnification x2.5).

Fig 2: Comparison between the strain and substrate elemental composition

Fig 3: Evolution of the WB mass balance in both insoluble and permeate fractions regarding granulometry

Fig 4: Evolution of WB elemental composition for (A) class 2 and (B) class 3 in both insoluble and permeate fraction

Fig 5: Evolution of total, organic and mineral nitrogen in both insoluble (A) and permeate (B) fractions among all WB classes 2, 3 and 4.

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